COURSE OUTLINE

(1) GENERAL

SCHOOL	School of Engineering					
ACADEMIC UNIT	Department of Financial and Management Engineering					
LEVEL OF STUDIES	Undergraduate					
COURSE CODE	MH0106	SEMESTER 4 th				
COURSE TITLE	Chemical Technology					
INDEPENDENT TEACHING ACTIVITIES if credits are awarded for separate components of the course, e.g. lectures, laboratory exercises, etc. If the credits are awarded for the whole of the course, give the weekly teaching hours and the total credits			WEEKLY TEACHING HOURS		CREDITS	
			3		4.5	
Add rows if necessary. The organisation of teaching and the teaching methods used are described in detail at (d).						
COURSE TYPE general background, special background, specialised general knowledge, skills development	General bac	kground				
PREREQUISITE COURSES:	No					
LANGUAGE OF INSTRUCTION and EXAMINATIONS:	Greek					
IS THE COURSE OFFERED TO ERASMUS STUDENTS	No					
COURSE WEBSITE (URL)						

(2) LEARNING OUTCOMES

Learning outcomes

The course learning outcomes, specific knowledge, skills and competences of an appropriate level, which the students will acquire with the successful completion of the course are described.

Consult Appendix A

- Description of the level of learning outcomes for each qualifications cycle, according to the Qualifications Framework of the European Higher Education Area
- Descriptors for Levels 6, 7 & 8 of the European Qualifications Framework for Lifelong Learning and Appendix B
- Guidelines for writing Learning Outcomes
 - Main purpose of this course is to introduce students to issues such as unit systems, mass and energy balances, calculations in chemical technology, fluid flow, heat transfer, processing and use of water in the chemical industry, hygiene and safety in the chemical industry. After successful completing the course, students will be expected to be able to:
 - Understand the basic principles of physical and chemical processes used in the production process as well as the rules that govern
 - Understand processes such as catalysis, adsorption, precipitation and filtration

Gene	eral Competences
-	Understand issues regarding water technology
-	Understand the anti-pollution technology
-	Evaluate the efficiency of processes
	industrial instrument
-	Resolve application exercises in various systems of physical-chemical processes of

Taking into consideration the general competences that the degree-holder must acquire (as these appear in the Diploma Supplement and appear below), at which of the following does the course aim?

Search for, analysis and synthesis of data and information, with the use of the necessary technology Adapting to new situations Decision-making Project planning and management Respect for the natural environment Criticism and self-criticism

(3) SYLLABUS

There will be analysis of physical and chemical processes that occur in the production of goods and the understanding of the basic rules and principles that govern these processes. Particular attention is paid to certain processes, catalysis, adsorption, sedimentation and filtration. Examples and exercises of application in various industrial systems are given for the better understanding of the chemical and natural processes. Special topics on water technology are also presented.

Module Contents (Syllabus):

Introduction to Technical Calculations		
Basic definitions		
Mass Balances		
Energy Balances		
Chemical Equilibrium		
Intermediate Examination		
Chemical Kinetic		
Physicochemical Processes		
Heterogeneous Processes		
Catalysis		
Drinking water and wastewater technology		
Adsorption		
Precipitation		
Filtration		

(4) TEACHING and LEARNING METHODS - EVALUATION

DELIVERY Face-to-face, Distance learning, etc.	Face-to-face			
USE OF INFORMATION AND COMMUNICATIONS TECHNOLOGY	Use of ICT in teaching			
Use of ICT in teaching, laboratory education, communication with students				
TEACHING METHODS	Activity	Semester workload		
The manner and methods of teaching are	Lectures	39		
Lectures, seminars, laboratory practice, fieldwork study and analysis of	Study and analysis of	98		
bibliography, tutorials, placements, clinical	Intermediate	3		
educational visits, project, essay writing,	examination			
artistic creativity, etc.	Examinations	3		
The student's study hours for each learning activity are given as well as the hours of non-directed study according to the				
principles of the ECTS				
	Course total	143		
STUDENT PERFORMANCE EVALUATION Description of the evaluation procedure Language of evaluation, methods of evaluation, summative or conclusive, multiple choice questionnaires, short-answer questions, open-ended questions, problem solving, written work, essay/report, oral examination, public presentation, laboratory work, clinical examination of patient, art interpretation, other Specifically-defined evaluation criteria are given, and if and where they are accessible to students.	Intermediateand final exa the semester, which inclu (development), knowledg the content of the course	minations at the end of de questions se and understanding of , and problem solving.		

(5) ATTACHED BIBLIOGRAPHY

Suggested bibliography:

Βασικές Αρχές και Υπολογισμοί στη Χημική Μηχανική (D. M. Himmelblau, J.B. Riggs, 2006) Εισαγωγή στη Χημική Τεχνολογία (Ζαμπούλης, Δ., Ζουμπούλης Α., Μάτης Κ., Μαύρος, Π., 2009) Χημική Τεχνολογία (Ζαμπούλης, Δ., Ζουμπούλης Α., Καραπάντσιος Θ., Μάτης Κ., Τριανταφυλλίδης Κ., 2013) Βασικές Αρχές και Υπολογισμοί στη Χημική Μηχανική (D. M. Himmelblau, J.B.

Riggs, 2017) Μηχανική Χημικών Διεργασιών (J. M. Smith, 1997) Notes

AdditionalReferences:

Περιβαλλοντική Μηχανική Ι - Διαχείριση Υδατικών Πόρων (Θ. Λέκκας, 1996)

Χημική Ισορροπία (Σ. Ι. Καρέκου, 1997)

- Related academic journals:

Journal of Chemical Technology and Biotechnology

Chemical Engineering Journal